

## Chemistry F335 June 2013 Mark Scheme

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### Chemistry F335 June 2013 Mark

F335 Mark Scheme June 2013 4 1 (a) (iv) OCOHO H 1. hydrogen bond (shown as in diagram or by dashed line), with bond shown straight (less than 20 o angle) with H-O 2. partial charges as shown (for any one hydrogen bond) 3. lone pair on CO<sub>2</sub> oxygen pointing along bond (for any one hydrogen bond)

### Advanced GCE Unit F335: Chemistry by Design

F335 Mark Scheme June 2013 8 Question Answer Marks Guidance 3 (a) HC O O H 1 O-H must be displayed to gain mark shape is unimportant 3 (b) (i) CH<sub>2</sub>O<sub>2</sub> CHO<sub>2</sub> - + H<sup>+</sup> 1 ALLOW structural formulae for anion with atoms in any order and 'OO' for 'O<sub>2</sub>') Allow negative charge anywhere. ALLOW H<sub>2</sub>O added on LHS with H<sub>3</sub>O<sup>+</sup> on right IGNORE '(aq)', other state symbols are CON

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2 marks Advanced GCE Unit F335: Chemistry by Design June 2013 F335 Paper. Question paper -  
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F334 Mark Scheme June 2014 4 Question Answer Mark Guidance 2 a 2-hydroxypropanoic acid  
2-hydroxy propanoic acid 2 IGNORE 'dashes & commas' & space between 'hydroxy' and 'propanoic'  
acid DO NOT ALLOW propaneic acid, hydroxyl, hydroxo b moles of NaOH =  $1.00 \times 33.6 / 1000 =$   
 $0.0336$  moles of acid in  $25.0 \text{ cm}^3 = 0.0336$  ...

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