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Chapter 15. Surface Tension. Surfactant. Aqueous Solution. Solvent. An inward force that tends to minimize the surface area of a liquid. Any substance that interferes with the hydrogen bonding between water molecules. Water that contains dissolved substances.

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enthalpy of the reactants.

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chemistry chapter 15 vocabulary.  
surface tension. surfacant. aqueous solution. solvent. an inward force that tends to minimize the surface area of a l.... any substance that interferes with

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the hydrogen bonding between.... water that contains dissolved substances. the dissolving medium in a solution.

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and displaces 0.81 cm<sup>3</sup> of water in a graduated cylinder. What is the density of gold? a. 0.81 grams/centimeters cubed ... Chemistry Chapter 2

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21. Calculate the specific heat in J/(g C)



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of an. 1. Analyze each of the five regions of the graph, unknown substance if a 2.50 g sample releases 12.0 cal as its temperature changes from 25.0 C to 20.0 C.  $q = c m \Delta T$   $12 \text{ cal} = (c)(2.50 \text{ g})(5.0 \text{ C})$   $c = \frac{12 \text{ cal}}{(2.50 \text{ g})(5.0 \text{ C})} = 4.02 \text{ J/(g C)}$

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Answer Anions are attracted to the  
hydrogen atoms of the water molecule,  
because hydrogen atoms are slightly  
positive.

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bias in their execution and interpretation. For this reason, external methods of quality assessment also play an important role in quality assurance programs.

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for Quick Review; 6. The concentrations of the reactants, A and B, and the initial reaction rates (just after  $t = 0$ ) for the following reaction in three different experiments are reported in the table. Complete the expression for the rate law for this reaction and determine the rate constant,  $k$ .

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