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Use  $T_1 = 500 \text{ K}$ ,  $V_1 = 1 \text{ L}$ , and  $T_2 = 300 \text{ K}$   $T_1 = 500 \text{ K}$ ,  $V_1 = 1 \text{ L}$ , and  $T_2 = 300 \text{ K}$  for your plot. 78 . Two moles of a monatomic ideal gas such as helium is compressed adiabatically and reversibly from a state (3 atm, 5 L) to a state with pressure 4 atm. (a) Find the volume and temperature of the final state.

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## **Ch. 3 Problems - University Physics Volume 2 | OpenStax**

Introduction to Dynamics: Newton's Laws of Motion; 4.1

Development of Force Concept; 4.2 Newton's First Law of Motion: Inertia; 4.3 Newton's Second Law of Motion: Concept of a System; 4.4 Newton's Third Law of Motion: Symmetry in Forces; 4.5 Normal, Tension, and Other Examples of Forces; 4.6 Problem-Solving Strategies; 4.7 Further Applications of Newton's Laws of Motion

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